

プロジェクト名：結晶Si/有機ハイブリッド太陽電池の高効率化に関する研究
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Effects of molybdenum oxide molecular doping on the chemical structure of poly(3,4-ethylenedioxythiophene):poly(stylenesulfonate) and on carrier collection efficiency of c-Si/PEDOT:PSS heterojunction solar cells

The effects of MoO₃ molecular doping in poly(3,4-ethylenedioxythiophene):poly(stylenesulfonate) (PEDOT:PSS) on the chemical structure and, in turn, on the carrier collection efficiency of c-Si/PEDOT:PSS heterojunction solar cells are demonstrated. Scanning electron microscopy revealed that the hydrophilic PSS polymer chain was intercalated into the interlayer van der Waals gap of MoO₃ flake sheets, which modified the chemical structure of PEDOT:PSS. MoO₃ exhibited intense photoluminescence in the 350-550 nm region, which enhanced the carrier collection efficiency of c-Si/PEDOT:PSS heterojunction solar cells with no significant changes. These findings suggest that the intense photoluminescence of MoO₃ and its light wavelength conversion contribute to the increased carrier collection efficiency.

1. Introduction

There has been tremendous interest in developing next-generation photovoltaics based on advanced materials such as quantum dots, semiconducting nanostructures and conjugated polymers.¹⁻³ Crystalline silicon (c-Si) remains the basic material for high-efficiency stable photovoltaics, and it has recently been demonstrated that Si nanowire junctions and Si wire arrays from catalytic growth or electrochemical etching can improve light absorption.⁴⁻⁶ On the other hand, the integration of different nanostructures or polymers into Si-based photovoltaics has been explored widely, and many heterojunction structures have been reported, such as GaN nanorods epitaxially grown on n-type c-Si, InAs nanorods grown on p-type Si, and multiwalled nanotube (CNT) arrays grown in an anodized alumina template on Si substrates.⁷⁻⁹

Highly conductive poly(3,4-ethylenedioxythiophene):poly(stylenesulfonate) (PEDOT:PSS) is often used to modify indium tin oxide (ITO) electrodes owing to its superior injection/collection properties by secondary doping of novel reagents such as graphene, graphene oxide (GO), CNTs, and metal oxide particles.¹⁰⁻¹³ It is also reported that PEDOT:PSS, GO, and metal oxide such as molybdenum trioxide (MoO₃) are good candidates for solution-processable hole-transporting materials that exhibits an intense photoluminescence (PL) emission in the 325-550 nm region and is expected to find application as a light wavelength conversion layer.¹⁴⁻¹⁷ Furthermore, the MoO₃ layer would presumably generate a strong inversion in Si by electrostatic repulsion, as in the cases of dielectric SiO_x and Al₂O₃.¹⁸⁻²⁰ Thus, a high built-in voltage, V_{bi}, can be expected at the c-Si/PEDOT:PSS:MoO₃ composite interface.

In the present paper, we demonstrate the effect of luminescent MoO₃ molecular doping to conductive PEDOT:PSS on the chemical structure and external quantum efficiency (EQE) of n-type c-Si(100)/PEDOT:PSS heterojunction solar cells.

2. Experimental details

The c-Si wafer (CZ, 3-5 $\Omega\cdot\text{cm}$, 300 μm thickness) was cleaned by the RCA1 and RCA2 procedures. Highly conductive PEDOT:PSS (Clevios PH1000) with 5 wt% DMSO was used as a starting solution. MoO_3 (Kanto Kagaku Ltd. 99.99%) powder flake with an average size of 1.7 μm were used as a dopant. PEDOT:PSS: MoO_3 composite solutions with different $\text{MoO}_3/(\text{PEDOT:PSS}+\text{MoO}_3)$ (1, 3, 10, and 30 wt%) were stirred for 1h. As the PEDOT:PSS: MoO_3 solution was then stirred for 1h. Here, as the PEDOT:PSS solid content of PH1000 is around 1%,²¹ the dry weight ratios of $\text{MoO}_3/\text{PEDOT:PSS}$ composite thin films were much higher than the values reported in present paper, for example, in the case of the 1 wt% composite, the real dry weight ratio of $\text{MoO}_3/\text{PEDOT:PSS}$ was almost unity. A thin film was formed by spin-coating the precursor solution on RCA-cleaned c-Si wafers at 1000 rpm for 60 s, followed by thermal annealing at 140°C for 30 minutes. Finally, a top Ag grid electrode was fabricated by pattern printing. The PEDOT:PSS: MoO_3 composite films were characterized using scanning electron microscopy (SEM), energy dispersive X-ray spectroscopy (EDX), X-ray photoelectron spectroscopy (XPS), and spectroscopic ellipsometry (SE). The PL spectra were measured using FluoroMax-3 (Horiba Jovin-Yvon) spectrophotometer.

3. Results and discussion

Figure 1(a) shows the SEM image of MoO_3 particles in conductive PEDOT:PSS. The inset shows the crystal structure of MoO_3 . MoO_3 exhibits a layered structure consisting of vertex-sharing chains of distorted MoO_6 octahedra that share edges with two similar chains to form layers. The distortion gives rise to very short “Mo=O” bonds at the apical octahedral positions that terminate the top and bottom of the double-octahedral oxide sheets.

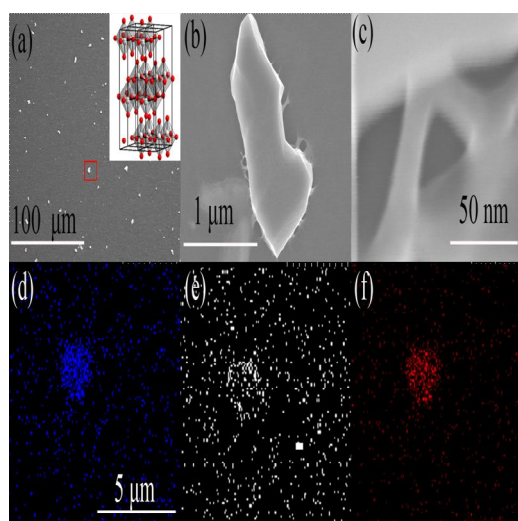


FIG 1 SEM images (a-c) and EDX maps (d-f) (corresponding to the circled region in Fig. 1 (a)) of MoO_3 particle in PEDOT:PSS. Inset illustrates crystal structure of MoO_3 where red and gray spheres denote O and Mo atoms, respectively, (b) and (c) show the novel hair-like structure. (d), (e), and (f) are the EDX maps for Mo, O, and S, respectively, in MoO_3 doped PEDOT:PSS.

The resultant two-dimensionally bonded layers are stacked in a staggered arrangement and held together by weak van der Waals forces. A variety of large guest species can be intercalated into the interlayer van der Waals gap.²² The SEM image shows a hairlike projections around the MoO_3 flake sheets with an average diameter of 15–20 nm which is much smaller than the typical mean size of PEDOT particles embedded in PSS chains.²³ EDX maps of Mo, O, and S for the PEDOT:PSS: MoO_3 (3 wt%) composite are shown in Figs. 1(d)-(f). Notably, the EDX image of Mo was almost consistent with that of S despite the use of MoO_3 , and even though S was distributed homogeneously. This implies that

the sulphur in PEDOT and/or PSS is coordinated preferentially in MoO₃ particle. In addition, the average *d* spacing of the collected composite powder, as calculated from the four *0k0* reflections in the XRD spectra, was 0.9-1.1 nm, i.e., larger than that of pristine MoO₃ (0.6 nm). Furthermore, XPS revealed that the 3d binding energy corresponding to S=O in PSS decreased less for higher MoO₃ concentrations. These findings suggest that the concentration ratio of PEDOT to PSS increased. Thus, the PSS chain was intercalated into the MoO₃ flake sheet, modifying the chemical structure of PEDOT:PSS conjugated polymer.

Figure 2 shows the PL spectra of pristine MoO₃ and PEDOT:PSS:MoO₃ composite films at room temperature and the excitation PL (PLE) spectra for the 400 nm emission. Optical micrographs of PEDOT:PSS with and without MoO₃ doping are shown as well. An intense broad band PL emission at 350-500 nm having a maximum peak at around 390-400 nm was observed, and its intensity increased with MoO₃ doping concentration. Thus, the UV light at 310-345 nm was absorbed and efficiently showed an intense broadband PL emission at ~400 nm. In fact, the MoO₃-doped PEDOT:PSS micrograph was deep purple, owing to the intense PL emission at 350-550 nm from MoO₃. These results suggest that MoO₃ may potentially be used for absorbing light at short wavelengths and then emitting at longer wavelengths (350-550 nm), which would enhance the carrier collection efficiency of c-Si/PEDOT:PSS solar cells.

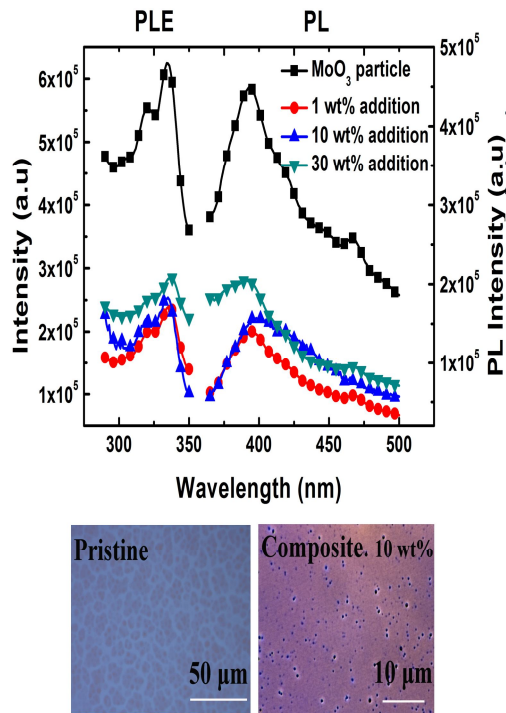


FIG 2 PL and PLE spectra of MoO₃ doped PEDOT:PSS film under 400-nm excitation. The optical micrographs of PEDOT:PSS with and without 10 wt% MoO₃ doping are shown as well.

Figure 3(a) shows the current density-voltage (*J*-*V*) curves obtained under AM1.5 100 mW/cm² simulated solar light (Bunkoukeiki CEP-25BX) for c-Si(100)/PEDOT:PSS:MoO₃ composite heterojunction solar cells with different MoO₃ doping concentrations. The inset is a schematic of a c-Si/organic heterojunction solar cell. The effective area of each cell was 5×5 mm². η was calculated via $\eta = V_{oc} J_{sc} FF / P_{in}$, where V_{oc} is the open-circuit voltage, J_{sc} is the short-circuit current density, FF is the fill factor, and P_{in} is the incident light power. FF is determined via $FF = (V_m J_m) / (V_{oc} J_{sc})$, where V_m and J_m are, respectively, the voltage and current density in the maximum power in the fourth quadrant of the *J*-*V* curves. The reverse and forward dark currents were suppressed to below the values for a pristine PEDOT:PSS diode. η increased to 11.01% with a J_{sc} of 29.4 mA/cm², a V_{oc} of 0.552 V, and an FF of 0.683 at a MoO₃ concentration of 1 wt%.

J_{sc} increased with MoO_3 doping concentration up to 3 wt%. However, V_{oc} and FF decreased markedly with further increase in MoO_3 concentration, due to the increased series resistance, R_s . The $1/C^2$ - V plot at 1 MHz also showed a relatively high V_{bi} of 0.9-1 eV, resulting in a higher V_{oc} than that of pristine PEDOT:PSS. The EQE and difference EQE (DEQE) spectra for solar cells with different MoO_3 doping concentrations are shown in Fig. 3(b). Here, the DEQE values are relative to that of a pristine c-Si/PEDOT:PSS solar cell, DEQE was found to be positive over the entire wavelength region of 350-1200 nm for a MoO_3 concentration of 1-3 wt%. At high MoO_3 doping concentrations of up to 10-30 wt%, DEQE was markedly enhanced in the 350-550 nm regions, with a shorter wavelength shift, but the values in the infrared region of 900-1200 nm were negative. It has been reported that polar solvent such as MeOH and ethylene glycol modify the chemical structure of PEDOT:PSS and that the phase separation of PEDOT:PSS enhances the hole mobility of PEDOT:PSS by increasing conduction path network of PEDOT chains.^{24,25} These findings suggest that the increased EQE in the 350-550 nm region of c-Si/PEDOT:PSS: MoO_3 composite heterojunction solar cells is due to the PL emission of MoO_3 .

The PLE and PL spectra of pristine MoO_3 are shown in Fig. 4(a). The EQE and DEQE spectra of c-Si/PEDOT:PSS solar cells with and without a 30 wt% MoO_3 doping and spectroscopic sensitivity of sunlight are also included. The PL spectra at 360-510 nm can be ascribed to the intense near-band-edge (NBE) emission due to the free-exciton recombination and visible light emission by transitions of excited optical centers in the deep levels (DLs). The UV absorption between 310-345 nm led to visible PL emission in the 360-510 nm range. This visible wavelength region corresponds to those of the increased DEQE for c-Si/PEDOT:PSS:GO composite solar cells.

The reflectance spectra for PEDOT:PSS: MoO_3 composite films, as determined by SE spectra analysis, are shown in Fig. 4(b). The reflectance spectra in the UV region corresponding to the absorption of c-Si decreased slightly by MoO_3 doping, but no significant changes were observed. AFM revealed that the degree of RMS roughness was 3-9 nm ($5 \times 5 \mu\text{m}^2$), which is less than that of light trapping. Thus, the improved EQE in the

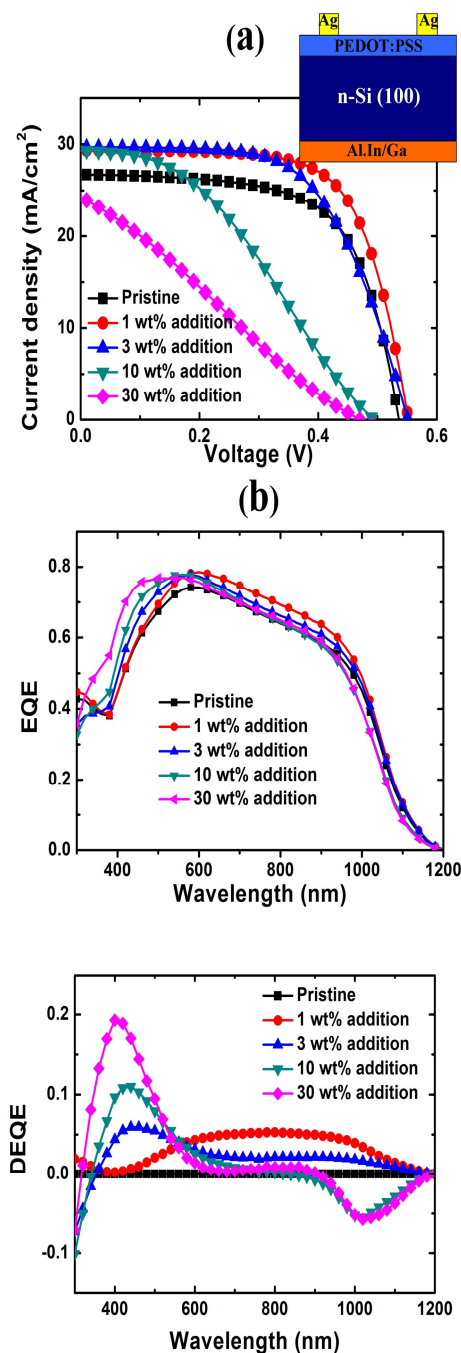


FIG 3 (a): J - V curves for n-Si/PEDOT:PSS: MoO_3 composite heterojunction solar cells with different MoO_3 concentrations. Inset shows schematic of the device structure of c-Si/PEDOT:PSS: MoO_3 composite solar cells. (b): EQE and DEQE spectra for c-Si/PEDOT:PSS: MoO_3 composite heterojunction solar cells. DEQE values are relative to that of a pristine PEDOT:PSS solar cell.

350-550 nm region does not result from MoO₃ acting as an anti-reflection layer but from the enhanced optical absorption in this region owing to the PL emission from MoO₃. The RMS roughness of the thin film examined by AFM was around 9 nm, up to a MoO₃ doping concentration of 10 wt%, which is negligibly small for a light-harvesting structure, as compared with that of textured Si(100). However, the EQE in the infrared region of 900-1200 nm deteriorated at a MoO₃ doping concentration above 10 wt% due to the increased optical reflectivity.

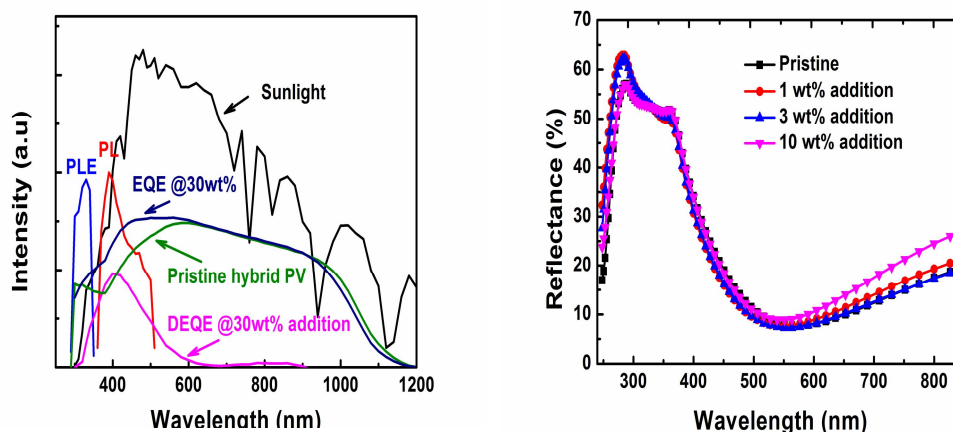


FIG 4 (a) The PLE and PL spectra of pristine MoO₃. The EQE, DEQE of c-Si/PEDOT:PSS solar cells and spectroscopic sensitivity of sunlight are also shown. (b) Reflectance spectra for c-Si(100)/PEDOT:PSS:MoO₃ composite solar cells with different MoO₃ doping concentrations. All PEDOT:PSS:MoO₃ layer thicknesses were 100 nm.

4. Conclusions

We studied the effect of MoO₃ molecular doping to conductive PEDOT:PSS on the chemical structure and photovoltaic properties of c-Si/PEDOT:PSS heterojunction solar cells. The MoO₃ molecular doping modified the chemical structure of PEDOT:PSS by the intercalation of PSS into the van der Waals gap of MoO₃ sheets. These changes enhanced the photovoltaic performance of the c-Si/PEDOT:PSS solar cells. η increased up to 11.0%. The UV absorption and intense PL emission at 350-550 nm increased the carrier collection efficiency in c-Si/PEDOT:PSS heterojunction solar cells.

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